Chemistry 24 Syllabus - Dr. Kline - Spring 2014

Lecture/Lab: Tu 2:00-7:05 p.m. or Th 5:30-10:35 p.m., Sci 305

Dr. Kline Contact Information
- Office: Sci 272
- E-mail: kline_peggy@gapps.smc.edu; kline_peggy@smc.edu (the second one forwards to the first one, so either is OK)
- Phone: 310-434-4745
- Web Site: homepage.smc.edu/kline_peggy/

Office Hours
MW 5-6 pm, Tu 12:45-1:45 pm, and Th 3:30-4:30 pm; also online via eCompanion (access via Corsair Connect; will usually get response within 24-36 hours)

Books, Supplies, and Internet

Books and Supplies
- Techniques in Organic Chemistry, 3rd or 2nd ed., by Mohrig, Hammond, Schatz, and Morrill - required
- Lab Experiments (Procedures) - SMC Chem 24 - Organic Chem II Laboratory Manual. This consists of plastic-wrapped loose-leaf pages taken from the book Modern Projects and Experiments in Organic Chemistry, 2nd ed. (Miniscale and Standard Taper Microscale), by Mohrig, Hammond, Schatz, and Morrill.; you may use the book itself which can be found online for less money than the loose-leaf pages and will work for Chem 21 and Chem 24.
- Lab Goggles. Locker Card, and Lab Notebook (lined pages that are not easily removed) - all required
- Nitrile or Neoprene Gloves – required
- Blue Flame-Retardant Lab Coat - required

Available via Internet
- Class Web Page (http://homepage.smc.edu/kline_peggy/chemistry-24---organic/)
- Class Bulletin Board - eCompanion
- Turnitin.com - submit lab reports electronically

Grading

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<th>Assessment</th>
<th>Points</th>
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<td>Tests (3)</td>
<td>180</td>
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<td>Lab Reports (8 of 9)</td>
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<td>Qualitative Analysis</td>
<td>20</td>
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<td>Total =</td>
<td>280</td>
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Standards (based on total points):
- A ≥ 252 points (90%)
- B ≥ 224 points (80%)
- C ≥ 182 points (65%)
- D ≥ 140 points (50%)
- F < 140 points

Lab Tests
There are three lab exams, worth 60 points each. They will be given during the lab time and will take 60-90 minutes. You may or may not be allowed use your personal lab notebook during the exam; however, no photocopied material is allowed in your notebook, with the exception of, spectral reference tables, other reference material (pH paper, heating mantle information, etc.) and spectra. All of these must be firmly affixed on all sides and not flapping. See the Lab Notebooks and Reports handout for more information. Topics for lab exam questions include questions answered for the lab reports, assigned reading, pre-lab lectures, and the experience of performing and writing up the labs. There will be no makeup tests or quizzes.
Lab Reports
The lab report for each experiment consists of a typed portion (submitted via Turnitin) and the notebook pages for that experiment (appended to the Turnitin copy or submitted as hard copies). See the Lab Notebooks and Reports handout for more information. The due date is usually the lab following the one when the experiment is completed. Grades are based on your answers to the questions, the organization and planning of your notebook, your analysis and results, and your conclusion. Do not commit plagiarism by copying the lab manual, another book, someone else's report, something you find online, or anything else. Plagiarism is a form of cheating and will not be allowed. Both the source and the recipient may be penalized. Late labs will be penalized at the rate of -2 points per commenced week.

Notebooks. You must write in pen in a blank, bound laboratory notebook. Notebooks may be evaluated “informally” throughout the semester, as well as formally grading your notebook copies with each experiment report. Deductions result from omissions, “fudging” of data, incoherence, lack of format, and other reasons. Deductions may occur at any point during the semester, and will occur if the instructor observes data being written anywhere except in your notebook. The instructor reserves the right to collect your notebook at any time during the semester. Again, there are more details in the Laboratory Notebooks and Reports handout.

Class Information
• Course material will be posted on or linked from the class web site: http://homepage.smc.edu/kline_peggy/chemistry-24---organic/. The instructor will send out communications to students using their official SMC email addresses and/or eCompanion so make sure you check the email addresses associated with both of those. You are responsible for knowing about information sent to your official SMC address and via eCompanion.
• Contacting the instructor. Use the Threaded Discussion (Q&A) area of eCompanion to ask course-related (content and protocols) questions. Use email only for personal questions. Include the course name as the subject in any email to help keep it from getting trapped in the spam filter. The instructor reserves the right to ignore questions asked via email that should have been asked via the threaded discussion.
• Successful completion of this course will require full participation in all class activities. Punctuality is critical as well—plan to arrive on time each and every class period. You will miss important material, annoy your fellow students and anger the instructor when you disrupt the class by entering late. Students are responsible for knowing what happens in class, including schedule changes, material not in the book, information about what’s going to be on the next test and so on. It is a good idea to have the names and contact information for a few students whom you can contact if you miss class.
• Electronic devices. Please adjust cell phones, laptops, and tablets, etc. so they do not make noise and/or disrupt class members; the instructor reserves the right to confiscate such devices that do make noise and/or to evict students who are not using them appropriately during class time.
• No eating, gum chewing, or drinking is permitted in classrooms or labs; no food or drink is permitted unless it’s sealed so that it absolutely cannot spill.
• Tutoring. The Science Learning Resource Center (Sci 245, http://www.smc.edu/sciencelrc/) provides free tutoring for SMC students and other resources; see Saundra Willis (434-4630) to set up a tutoring appointment.
• Religious Holiday Absences. SMC Academic Regulation 5530 states: “It is the college practice that students may be required to make-up missed work from absences due to the observance of a religious holiday, but they cannot be penalized for such absences. This practice applies to any work affecting a student’s grade.” My policy is to avoid scheduling tests and quizzes on religious holidays that commonly affect students in my classes and, whenever possible, to schedule labs that can easily be rescheduled or done on a student’s own time on religious holidays that affect large numbers of students. Students must let the instructor know by email within the first week of class of any planned absence due to a religious holiday. I try to offer students opportunities to make up labs they have to miss and expect students to make a reasonable effort to make up said labs. I do lecture on religious holidays and students who need to miss class are expected to get notes from other students.
• SMC accommodates students with disabilities. If you qualify for any special accommodations due to a disability, you need to officially process your request through the Disabled Students Programs and Services (DSPS) office. If you believe you have a learning disability that has not yet been documented, please see me and make an appointment at the DSPS office for assistance. The DSPS office is located...
in the Admissions/Student Services Complex, Room 101, and the phone numbers are (310) 434-4265 and (310) 434-4273 (TDD). Students requiring permissible accommodations should contact the instructor by email no later than the end of the second week or classes or as soon as s/he becomes aware of the disability. No retroactive accommodations will be provided. The student is solely responsible for securing any provisions to which they may be entitled. Scheduling of accommodated exams must be made through DSPS.

- **Missed labs.** There will be no make-up labs, unless you arrange to do the lab during another lab time when that section is doing the experiment. If you miss one lab, it will be the one dropped. If you miss more than one lab, you will forfeit the points. Students who must miss their regularly-scheduled lab time due to a day of religious observance may not be penalized; however, they must make a reasonable attempt to attend another session and/or complete the lab report.

- **There will be no makeup tests or quizzes.** If you miss a test for a legitimate, verifiable reason the normalized score on the final will be used in place of that test score.

- **“Graphing” calculators** are permitted during tests for which calculators are permitted until the first occurrence of one being used improperly; from that point on, they will not be permitted for any student in any class. No cell phones, dictionaries, or translators are allowed during quizzes or tests.

- **The Academic Honesty Policy of Santa Monica College** will be strictly enforced. Acts of academic dishonesty including, but not limited to, plagiarism, providing test/quiz answers to another student, and copying from another student can result in a failing grade for the assignment or the course. Plagiarism consists of presenting the words of another person as your own and includes “recycling” written work from other students and the Internet. Both the provider and the recipient of the information will be penalized. In addition, lying, manipulative or disruptive behavior will not be tolerated. More information on SMC policies is available on the website for The Office of Student Judicial Affairs. Students need to be familiar with the SMC Code of Academic Conduct.

- **Re-grading.** Tests may be submitted for re-grading (or re-adding) within one week of their initial return to students. Please note that the instructor reserves the right to re-grade the entire test or quiz. Answers that look as if they could have been changed after they were graded will not be considered for re-grading.

- **Turnitin.** This course will require electronic submission of written assignments through Turnitin (http://www.turnitin.com). Turnitin’s OriginalityCheck conducts textual similarity reviews of submitted papers. When papers are submitted to Turnitin, the service may retain a copy of the submitted work in the Turnitin database for the sole purpose of detecting plagiarism in future submitted works. Students retain copyright on their original course work. Please note that Turnitin does not accuse you of plagiarism; it is only identifying similarity to content in its database, and you or I will determine whether you used source material accurately and ethically. Additionally I will use the GradeMark feature to grade your labs online. The use of Turnitin is subject to the Terms of Use agreement posted on the Turnitin.com website. There is a Turnitin handout available in the Admissions/Student Services Complex, Room 101, and the phone numbers are (310) 434-4265 and (310) 434-4273 (TDD). Students requiring permissible accommodations should contact the instructor by email no later than the end of the second week or classes or as soon as s/he becomes aware of the disability. No retroactive accommodations will be provided. The student is solely responsible for securing any provisions to which they may be entitled. Scheduling of accommodated exams must be made through DSPS.

- **Dates and Deadlines.** See the SMC Dates and Deadlines web page for enrollment and payment deadlines and Corsair Connect for individual course withdrawal deadlines. The instructor reserves the right to drop any student who misses any class meetings during the first week. The instructor will probably drop students who miss a test without notification or appear to vanish; however, clerical errors do occur. If you want to be sure you are dropped, do it yourself. Aside from the circumstances under which you may be dropped by the instructor, it is nevertheless your responsibility as a student to withdraw from class if you do not intend to complete it. Students must not expect faculty to initiate withdrawal procedures for them. If you wish to drop this class, you may do so through Corsair Connect. Students may process a drop for themselves through 75% of the class, which is through the 12th week in a regular semester. Data regarding the withdrawal parameters for each class are provided within each student’s individual Corsair Connect account.

- **Safety, Personal Protective Equipment (PPE) and Emergency Information.** Chemical splash goggles must be worn by all students whenever they, or anyone else, are working with reagents in the laboratory. An extremely limited supply of goggles may be available in the lab for students to borrow. If you do not have a pair of appropriate goggles to wear for lab and none are available in the lab, you will have two options—go buy some or don’t work in lab that day. Students must supply nitrile or neoprene gloves for lab. Gloves must be removed and placed in the glove waste container when contaminated or any time you leave the lab. Students in organic chemistry classes must wear a blue flame-resistant lab coat whenever anyone in the room is conducting an experiment. If a student forgets his/her lab coat,
he or she may not remain in the lab without it. A limited supply of lab coats is available to rent from the
stockroom. A link to the SMC Safety Rules, PPE Information, and Emergency Information is on the
course web site. All students must sign a statement indicating that they are familiar with the above
Safety, PPE, and Emergency information before being allowed to work in the lab.

- **Lab neatness.** Students are responsible for keeping the lab neat. Each student will be assigned one
clean-up day during the semester. Any chemical samples kept at the end of the lab period must be
labeled with the student's and instructor's name, date, and identity of substance(s) present. Dispose of
unknowns in the appropriate waste container, clean the vial in which they came, and return the empty
clean vial to the instructor.

- **Lab Tests** may or may not be open lab notebook.

- **Additional lab information.** See the Lab Notebooks and Reports handout and the Chem 24 Lab
webpage (http://homepage.smc.edu/kline_peggy/chemistry-24-organic/) for additional information
about the lab and lab reports.

**Official Course Information**
Link to official course outline -
http://www.curricunet.com/SantaMonica/reports/course_outline_html.cfm?courses_id=242

**Catalogue Description**
This course is the second semester of organic chemistry laboratory. The laboratory work involves
synthesis, structure determination, reaction mechanisms, and qualitative analysis. The lectures will
discuss the theory and techniques that relate to the experiments that are performed, including NMR, IR,
organic qualitative analysis, and various forms of chromatography. A special emphasis will be placed on
FT-NMR, utilizing SMC’s 350 MHz spectrometer. Chem 21, 22, and 24 constitute two semesters of organic
chemistry with two semesters of laboratory. **Prerequisite: Chemistry 21 (Organic Chemistry I with Lab)
with a grade of C or better. It is imperative that students have previously taken an Organic Chemistry Lab.**

**Content**
Preparation of an n-Butyl Benzoate using a Phase Transfer Catalyst; Electrophilic Aromatic Substitution
Reaction; Synthesis of Para Red via a Diazonium Salt; Synthesis and Identification of an Ester from an
Unknown Alcohol; Preparation of An Amide of an Amino Acid; Synthesis and Using a Grignard Reagent;
Oxidation-Reduction Cycle of t-Butylcyclohexanol and t-Butylcyclohexanone; Aldol Condensation Reaction;
Organic Qualitative Analysis; NMR Analysis of Products—Including introduction to 2D NMR and
determination of a stereoisomeric mixture through analysis of complex splitting patterns in proton NMR.

**Student Learning Outcomes** As assessed by: questions on exams and/or observation of laboratory
performance and/or evaluation of notebook data and lab reports. 1-The student will follow a logical
process based on well-established scientific principles and demonstrate the ability to use the appropriate
problem-solving techniques to solve a scientific problem such as the determination of the structure of a
compound based on spectroscopy (IR, NMR, MS) and/or chemical evidence, or the prediction of a
compound’s chemical and/or physical behavior based on the behaviors of similar compounds. 2-When
conducting a laboratory experiment, the student will follow written procedures commonly used in the
organic lab (such as thin-layer chromatography, recrystallization and reflux) accurately and safely. The
student will maintain an accurate and organized lab notebook. When completing a lab report the student
will apply the scientific method correctly by being able to state a hypothesis, take careful measurements,
estimate uncertainties and draw appropriate conclusions based on gathered data and scientific principles.
3-The student will explain observable phenomena using appropriate scientific theories, such as explaining
the likely meaning of a lower-than-expected melting point, correlating the color and visible spectrum of a
molecule, or other observations made during lab experiments.

*Last revised 2/18/2014*