Chem 21 Syllabus - Winter 2015 - Dr. Kline

Meeting Times
Lecture: TuTh 1:30-6:40 pm in Sci 140 and F 1:30-6:40 pm in Sci 157
Lab/Lecture: MW 1:30-6:40 pm in Sci 305

Office Hours:
No official on-ground office hours, but may be available for about 30 minutes before class; online via eCompanion (access via Corsair Connect; will usually get response within 24-36 hours)

Dr. Kline Contact Information
• Office: Sci 272
• E-mail: kline_peggy@gapps.smc.edu; kline_peggy@smc.edu (the second one forwards to the first one, so either is OK)
• Phone: 310-434-4745
• Web Site: homepage.smc.edu/kline_peggy/

Grading

<table>
<thead>
<tr>
<th>Assessment</th>
<th>Points</th>
<th>Grade Standards (based on total points):</th>
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</thead>
<tbody>
<tr>
<td>Tests (2/3)</td>
<td>240</td>
<td>A ≥ 540 points (90%)</td>
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<tr>
<td>Quizzes (18/20)</td>
<td>90</td>
<td>B ≥ 480 points (80%)</td>
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<tr>
<td>Exercises (6)</td>
<td>30</td>
<td>C ≥ 390 points (65%)</td>
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<tr>
<td>Lab Reports (8/9)</td>
<td>80</td>
<td>D ≥ 300 points (50%)</td>
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<tr>
<td>Lab Citizenship</td>
<td>10</td>
<td>F &lt; 300 points</td>
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<tr>
<td>Lab Tests (3)</td>
<td>30</td>
<td></td>
</tr>
<tr>
<td>Final Exam</td>
<td>120</td>
<td></td>
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<tr>
<td>Total</td>
<td>600</td>
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The lower of the lowest test score or the final will be dropped.

Course Materials
Books and Supplies - all items are required unless otherwise noted
• Organic Chemistry, 2nd ed., by David Klein. The package sold in the SMC Bookstore includes a loose-leaf text, solutions manual, and a registration code for WileyPLUS. You can buy the same package directly from Wiley for a bit less money by following the link on the Chem 21 web page. Advice on using other texts: The 1st edition seems similar with respect to the organization of topics and many of the problems (but not necessarily the problem numbers) so you can probably make that work with some effort on your part. Using the Bruice text is going to require a lot of effort on your part as the organization is significantly different and the problems are obviously different. The instructor will not provide book section numbers for tests and quizzes for anything but the 2nd edition of Klein. The instructor does have problem set suggestions from the last few editions of Bruice.
• Study Guide and Solutions Manual to accompany Klein text – optional, but highly recommended. In addition to solutions for all of the problems, including in-chapter ones it includes the following: the review information at the back of each chapter in a workbook format, mistakes to avoid, and lists of useful reagents.
• **Molecular Model kit** – *optional, but highly recommended.* If you buy one on your own, either the *ball and stick* or “jacks” type (*Darling models*) are OK and each has its strengths and weaknesses. Both are also sold in the SMC bookstore. The ball and stick version sold in the bookstore has more carbon atoms (15 vs 6) than the one linked here.


• **Lab Techniques** - *Techniques in Organic Chemistry,* 4th ed., by Mohrig, Alberg, Hormeister, Schatz, and Hammond - (the 2nd or 3rd editions will work if you put in some effort; the 1st edition won’t)

• **Additional Lab Supplies:** Goggles. Nitrile or Neoprene Gloves, Locker Card, *blue flame-resistant Lab Coat,* and Lab Notebook (must be capable of making copies in real time).

**Internet Content**

- Copies of Old Tests and Quizzes - [http://homepage.smc.edu/kline_peggy/Exam_Archive/Chem21/default.html](http://homepage.smc.edu/kline_peggy/Exam_Archive/Chem21/default.html)
- Class Bulletin Board - in eCompanion (via Corsair Connect) - threaded discussion for Q&A
- **WileyPLUS** for Klein book. No online homework will be assigned for winter 2015 so you don’t need it now; however, you may need it for your Chem 22 course. The registration code sold in the bundled text in the SMC bookstore includes the online text (but not the solutions book), homework problems, study aids, and other items. This can also be purchased stand-alone; however, it’s pretty pricey (> $100). Note that you will need to create a WileyPLUS account, if you don’t already have one, first. The direct URL for our class is [http://edugen.wileyplus.com/edugen/class/cl392760/](http://edugen.wileyplus.com/edugen/class/cl392760/). The URL for the main WileyPLUS page is [https://www.wileyplus.com/WileyCDA/](https://www.wileyplus.com/WileyCDA/).

• **Mohrig Lab Book Websites:** *Techniques* (includes helpful videos!) - [http://bcs.whfreeman.com/mohrig4e/default.asp#t_923302](http://bcs.whfreeman.com/mohrig4e/default.asp#t_923302) **Projects** (what loose leaf pages are from) - [http://bcs.whfreeman.com/mohrig/default.asp?](http://bcs.whfreeman.com/mohrig/default.asp?)

**Course Information**

• **Course material** will be posted on or linked from the class web site: [homepage.smc.edu/kline_peggy/chem-21.html](http://homepage.smc.edu/kline_peggy/chem-21.html). The instructor will send out communications to students using their official SMC email addresses and/or eCompanion so make sure you check the email addresses associated with both of those. You are responsible for knowing about information sent to your official SMC address and via eCompanion.

• **Contacting the instructor.** Use the Threaded Discussion (Q&A) area of eCompanion to ask course-related (content and protocols) questions. Use email only for personal questions. Include the course name as the subject in any email to help keep it from getting trapped in the spam filter. The instructor reserves the right to ignore questions asked via email that should have been asked via the threaded discussion.

• Successful completion of this course will require full participation in all class activities. Punctuality is critical as well—plan to arrive on time each and every class period. You will miss important material, annoy your fellow students and anger the instructor when you disrupt the class by entering late. Students are responsible for knowing what happens in class, including schedule changes, material not in the book, information about what’s going to be on the next test and so on. It is a good idea to have the names and contact information for a few students whom you can contact if you miss class.
• **Electronic devices.** Please adjust cell phones, laptops, and tablets, etc. so they do not make noise and/or disrupt class members; the instructor reserves the right to confiscate such devices that do make noise and/or to evict students who are not using them appropriately during class time.

• **No eating, gum chewing, or drinking** is permitted in classrooms or labs; no food or drink is permitted unless it is sealed so that it absolutely cannot spill.

• **Tutoring.** The Science Learning Resource Center (Sci 245, http://www.smc.edu/sciencelrc/) provides free tutoring for SMC students and other resources; see Saundra Willis (434-4630) to set up a tutoring appointment.

• **Religious Holiday Absences.** SMC Academic Regulation 5530 states: “It is the college practice that students may be required to make-up missed work from absences due to the observance of a religious holiday, but they cannot be penalized for such absences. This practice applies to any work affecting a student’s grade.” My policy is to avoid scheduling tests and quizzes on religious holidays that commonly affect students in my classes and, whenever possible, to schedule labs that can easily be rescheduled or done on a student’s own time on religious holidays that affect large numbers of students. Students must let the instructor know by email within the first week of class of any planned absence due to a religious holiday. I try to offer students opportunities to make up labs they have to miss and expect students to make a reasonable effort to make up said labs. I do lecture on religious holidays and students who need to miss class are expected to get notes from other students.

• **SMC accommodates students with disabilities.** If you qualify for any special accommodations due to a disability, you need to officially process your request through the Disabled Students Programs and Services (DSPS) office. If you believe you have a learning disability that has not yet been documented, please see me and make an appointment at the DSPS office for assistance. The DSPS office is located in the Admissions/Student Services Complex, Room 101, and the phone numbers are (310) 434-4265 and (310) 434-4273 (TDD). Students requiring permissible accommodations should contact the instructor by email no later than the end of the second week or classes or as soon as s/he becomes aware of the disability. No retroactive accommodations will be provided. The student is solely responsible for securing any provisions to which they may be entitled. Scheduling of accommodated exams must be made through DSPS.

• **Topics for test and quiz questions** will be taken from information presented in lecture and assigned problems. Quizzes and Tests will be given at the beginning of the class period.

• **There will be no makeup tests or quizzes.** If you miss a quiz or test it will be dropped (up to two quizzes or one test only).

• “**Graphing**” calculators are permitted during tests for which calculators are permitted until the first occurrence of one being used improperly; from that point on, they will not be permitted for any student in any class. No cell phones, dictionaries, or translators are allowed during quizzes or tests.

• **Online homework** - no online homework for Winter 2015.

• **Suggested problems** for students to work on their own from the textbook are listed online: http://homepage.smc.edu/kline_peggy/chem-21/suggested-problems-from-the.html

• **The Academic Honesty Policy of Santa Monica College** will be strictly enforced. Acts of academic dishonesty including, but not limited to, plagiarism, providing test/quiz answers to another student, and copying from another student can result in a failing grade for the assignment or the course. Plagiarism consists of presenting the words of another person as your own and includes “recycling” written work from other students and the Internet. Both the provider and the recipient of the information will be penalized. In addition, lying, manipulative or disruptive behavior will not be tolerated. More information on SMC policies is available on the website for The Office of Student Judicial Affairs. Students need to be familiar with the SMC Code of Academic Conduct.

• **Re-grading.** Tests and quizzes may be submitted for re-grading (or re-adding) within one week of their initial return to students. Please note that the instructor reserves the right to re-grade the entire test or quiz. Answers that look as if they could have been changed after they were graded will not be considered for re-grading.

• **Turnitin.** This course will require electronic submission of written assignments through Turnitin (http://www.turnitin.com). Turnitin’s OriginalityCheck conducts textual similarity reviews of
submitted papers. When papers are submitted to Turnitin, the service may retain a copy of the submitted work in the Turnitin database for the sole purpose of detecting plagiarism in future submitted works. Students retain copyright on their original course work. Please note that Turnitin does not accuse you of plagiarism; it is only identifying similarity to content in its database, and you or I will determine whether you used source material accurately and ethically. Additionally I will use the GradeMark and Rubric features to grade your labs online. The use of Turnitin is subject to the Terms of Use agreement posted on the Turnitin.com website. There is a Turnitin handout from the instructor also linked from the navigation bar on the left side of the Chem 21 lab web page.

- **Safety, Personal Protective Equipment (PPE) and Emergency Information.** Chemical splash goggles must be worn by all students whenever they, or anyone else, are working with reagents in the laboratory. An extremely limited supply of goggles may be available in the lab for students to borrow. If you do not have a pair of appropriate goggles to wear for lab and none are available in the lab, you will have two options—go buy some or don’t work in lab that day. Students must supply nitrile or neoprene gloves for lab. Gloves must be removed and placed in the glove waste container when contaminated or any time you leave the lab. Students in organic chemistry classes must wear a blue flame-resistant lab coat whenever anyone in the room is conducting an experiment. If a student forgets his/her lab coat, he or she may not remain in the lab without it. A limited supply of lab coats is available to rent from the stockroom. A link to the SMC Safety Rules, PPE Information, and Emergency Information is on the course web site. All students must sign a statement indicating that they are familiar with the above Safety, PPE, and Emergency information before being allowed to work in the lab.

- **Lab neatness.** Students are responsible for keeping the lab neat. Students will be assigned one clean-up day during the semester. Any chemical samples kept at the end of the lab period must be labeled with the student’s and instructor’s name, date, and identity of substance(s) present. Dispose of unknowns in the appropriate waste container, clean the vial in which they came, and return the empty clean vial to the instructor. There are ten points available for lab citizenship: deductions from the ten points will occur for transgressions such as bottles left off lids, messy balance areas, not doing your assigned cleanup, not wearing appropriate PPE, etc. The entire lab class may get points deducted for certain items.

- **Missed labs.** There will be no make-up labs. If you miss one lab it will be the one dropped. Students who are absent for three or more labs and/or do not turn in three or more lab reports will receive a grade of F (or W, if they drop the course in time).

- **Lab Tests** may or may not be open lab notebook.

- **Additional lab information.** See the Lab Notebooks and Reports handout and the Chem 21 Lab webpage (http://homepage.smc.edu/kline_peggy/chem-21-lab.html) for additional information about the lab and lab reports.

- **Dates and Deadlines.** See the SMC Dates and Deadlines web page for enrollment and payment deadlines and Corsair Connect for individual course withdrawal deadlines. The instructor reserves the right to drop any student who misses any class meetings during the first week. The instructor will probably drop students who miss a test without notification or appear to vanish; however, clerical errors do occur. If you want to be sure you are dropped, do it yourself. Aside from the circumstances under which you may be dropped by the instructor, it is nevertheless your responsibility as a student to withdraw from class if you do not intend to complete it. Students must not expect faculty to initiate withdrawal procedures for them. If you wish to drop this class, you may do so through Corsair Connect. Students may process a drop for themselves through 75% of the class, which is through the 12th week in a regular semester. Data regarding the withdrawal parameters for each class are provided within each student’s individual Corsair Connect account.

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**Study Advice**

Chemistry is not a spectator sport—reading the book and watching the instructor work problems will not be sufficient. The only way to learn to work problems is by working them yourself (without looking at the solutions). Read the appropriate chapter sections and work the problems as we cover them in lecture. Do not wait until the last minute to work the problems. It is often helpful at least
to skim the chapter (read the headlines) before we cover it in lecture. If you have difficulty with a
problem, try rereading the appropriate section(s) or your lecture notes and/or the book, looking at
worked-out problems of the same type, etc. If you get totally frustrated with a problem, don’t get
hung up on it—go on to something else for a while. **Study strategies** that students have found
useful in the past include study groups, flashcards, working questions from old tests and quizzes,
reworking assigned problems and examples, and reviewing their tests before the final exam.

Classes taken during a short session will move at a rapid and, at times, alarming pace. Be
prepared!

**Official Course Information**

**Catalogue Description:** This course is a systematic introduction to the chemistry of carbon
compounds. It encompasses theory and reactions of hydrocarbons and functional group
derivatives. Included are bonding and structure, nomenclature, stereochemistry, synthesis,
mechanism, and spectroscopic analysis. The laboratory work focuses on techniques of synthesis,
-isolation, purification, and analysis. Chem 21 and 22 together constitute two semesters of organic
chemistry with one semester of organic chemistry laboratory. Chem 21, 22, and 24 constitute two
semesters of organic chemistry with two semesters of laboratory. **Prerequisite:** Chemistry 12 or
equivalent with a grade of C or better). Successful completion of this course requires access to a
computer with internet access to complete homework assignments, download handouts, obtain lab
information, and turn in lab reports.

**Content:** Review of Pertinent Chemistry 11 and 12 Topics—Bonding, Molecular Structure, Kinetics,
Equilibrium, and Acid-Base Chemistry; Introduction to Organic Compounds—Functional Groups,
Nomenclature, Reaction Concepts; Alkenes and their Reactions; Alkynes and their Reactions;
Stereochemistry; Spectroscopy (MS, IR, NMR, uV-vis); Electron Delocalization and Resonance;
Reactions of Dienes; Reactions of Alkanes; Nucleophilic Substitution and Elimination Reactions of
Alkyl Halides; Alkane Reactions; Laboratory Experiments—Experiments utilizing common organic
lab techniques such as reflux, distillation, extraction, recrystallization, melting point determination,
chromatography, and spectroscopy in the preparation, isolation, purification, and identification of
organic compounds.

**Student Learning Outcomes** As assessed by: questions on exams and/or observation of
laboratory performance and/or evaluation of notebook data and lab reports and/or lab tests.

1. The student will follow a logical process based on well-established scientific principles and
demonstrate the ability to use the appropriate problem-solving techniques to solve a scientific
problem such as an organic synthesis comprised of two or more steps, or a determination of
the structure of a compound based on spectroscopy (IR, NMR, MS) and/or chemical evidence.
2. When conducting a laboratory experiment, the student will follow written procedures commonly
used in the organic lab (such as reflux, distillation, extraction, recrystallization, and melting-
point determination) accurately and safely. When completing a lab report, the student will apply
the scientific method correctly by being able to state a hypothesis, take careful measurements,
estimate uncertainties and draw appropriate conclusions based on gathered data and scientific
principles.
3. The student will explain observable phenomena using appropriate scientific theories, such as
writing a reaction mechanism consistent with observed facts or utilizing appropriate structures
to represent different types of organic molecules and species.