Chem 21 Syllabus - Dr. Kline - Fall 2019

Class Times, Office Hours, Contact Info

Meeting Times
Lecture: MW 6:30-9 pm in Sci 157
Labs: Tu 2-6 pm or Th 6-10 pm in Sci 305

Office Hours
MW 4:30-6 pm; Th 4-5 pm or by appointment; also online
via Discussion area of Canvas (will usually get response
within 24-36 hours)

Web Site: homepage.smc.edu/kline_peggy/

Dr. Kline Contact Information
• Office: Sci 272
• E-mail: kline_peggy@gapps.smc.edu;
  kline_peggy@smc.edu (the most efficient way to
  contract this instructor)
• Phone: 310-434-4745 (inefficient method of contact)

Grading

<table>
<thead>
<tr>
<th>Required Activity</th>
<th>% of Course Grade</th>
<th>Comments</th>
</tr>
</thead>
<tbody>
<tr>
<td>Tests</td>
<td>44%</td>
<td>Three at 100 points each; none dropped</td>
</tr>
<tr>
<td>Lecture Quizzes</td>
<td>12%</td>
<td>Best 8/9 at 10 points each; lowest dropped</td>
</tr>
<tr>
<td>In-Class Activities</td>
<td>3%</td>
<td>About 7-8 in semester; 3 points each plus board points; up to 2 extra credit points overall possible</td>
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<tr>
<td>Labs</td>
<td>15%</td>
<td>Best 10 of 11 at 100 points each</td>
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<tr>
<td>Lab Quizzes</td>
<td>7%</td>
<td>Three at 15 points each; none dropped</td>
</tr>
<tr>
<td>Lab Citizenship</td>
<td>1%</td>
<td>Everyone starts with ten points; deductions taken.</td>
</tr>
<tr>
<td>Final Exam</td>
<td>18%</td>
<td>Cumulative. Sixty multiple-choice questions.</td>
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</tbody>
</table>

Grade Standards - based on overall weighted percentage

A ≥ 90.00 %
B ≥ 80.00%
C ≥ 65.00%
D ≥ 50.00 %
F < 50.00%

Testing, Grading and Academic Honesty

• Topics for test and quiz questions will be taken from information presented in lecture and/or assigned problems. Quizzes will be given at the end of the class period in a regular semester and at the beginning during a short session.
• No cell phones, dictionaries, or translators are allowed during quizzes or tests.
• There will be no makeup tests or quizzes. If you miss a test for a legitimate, documented reason your percent score on the final will be used in place of that test score. If you miss a quiz, it will be one that is dropped. If you miss more than two quizzes then you need to consult with the instructor.
• Re-grading. Tests may be submitted for re-grading (or re-adding) within one week of their initial return to students. Please note that the instructor reserves the right to re-grade the entire test or quiz. Answers that look as if they could have been changed after they were graded will not be considered for re-grading.

Regrades that involve questions scored by ZipGrade may be adjusted for all class members if not so doing creates an unfair disadvantage to students with one form of the test; this may result in students having a higher or lower score than originally scored.
• The Academic Honesty Policy of Santa Monica College will be strictly enforced. Acts of academic dishonesty including, but not limited to, plagiarism, providing test/quiz answers to another student, and copying from another student can result in a failing grade for the assignment or the course. Plagiarism consists of presenting the words of another person as your own and includes "recycling" written work from other students and the Internet. Both the provider and the recipient of the information will be penalized. In addition, lying, manipulative or disruptive behavior will not be tolerated. More information on SMC policies is available on the website for The Office of Student Judicial Affairs. Students must be familiar with the SMC Code of Academic Conduct.
Course Materials and Resources

Books and Supplies
- **Study Guide and Solutions Manual** to accompany Klein text – optional, but highly recommended. Has solutions for all of the problems, including in-chapter ones, the review information at the back of each chapter in a workbook format, mistakes to avoid, and lists of useful reagents.
- **Molecular Model kit** – optional, but highly recommended. If you buy one on your own, either the ball and stick or “jacks” type (Darling models) are OK and each has its strengths and weaknesses.
- **Lab Experiments (Procedures)** - available online: [http://homepage.smc.edu/kline_peggy/chem-21/lab/](http://homepage.smc.edu/kline_peggy/chem-21/lab/)
- **Lab Techniques** - Techniques in Organic Chemistry, 4th ed., by Mohrig, Alberg, Hormeister, Schatz, and Hammond
- **Lab Notebook** with NCR or carbon pages- required; the spiral-bound one sold in the SMC Bookstore on the shelves with the course textbooks is highly recommended.
- **Lab Goggles** (ANSI Z87.1)
- **Locker Card**
- **Nitrile or Neoprene Gloves**
- **Flame-Resistant Lab Coat** (Arc Rating ATPV 7.7 cal/cm²)

Available via Internet
- **Canvas** - [http://www.smc.edu/ACG/DistanceEducation/Canvas/Pages/Canvas-at-SMC.aspx](http://www.smc.edu/ACG/DistanceEducation/Canvas/Pages/Canvas-at-SMC.aspx) - includes grade book and threaded discussion. Labs will be submitted to Turnitin via Canvas.
- **WileyPLUS** Access at [www.wileyplus.com/class/599017](http://www.wileyplus.com/class/599017) for downloadable eText (incl. Pre-loaded content is there including eText (includes Study Guide), problems, video mini-lectures and solved problems, Reaction Explorer, ORION and more. Look under “Read, Study, Practice” for this content. The bookstore bundle includes a registration code. Not required.
- **Mohrig Website**: [http://www.macmillanlearning.com/Catalog/studentresources/mohrig4e#](http://www.macmillanlearning.com/Catalog/studentresources/mohrig4e#)
- **Lab Notebooks and Reports online handout** ([http://homepage.smc.edu/kline_peggy/chem-21/lab/lab-notebooks-and-reports.html](http://homepage.smc.edu/kline_peggy/chem-21/lab/lab-notebooks-and-reports.html)) - you are responsible for

Course Information

- **Course material** will be posted on or linked from the class web site: [http://homepage.smc.edu/kline_peggy/chem-21/](http://homepage.smc.edu/kline_peggy/chem-21/). The instructor will send out communications to students using their official SMC email addresses and/or Canvas so make sure you check the email addresses associated with both of those. You are responsible for knowing about information sent to your official SMC address and via Canvas.
- **Contacting the instructor**. Use the Discussion area of Canvas to ask course-related (content and protocols) questions. Use email only for personal questions. Include the course name as the subject in any email not sent via Canvas to help keep it from getting trapped in the spam filter.
- **Office hours**. These are for you. If there are already students in the office, please come in and join us. I take questions on a rotating basis if more than one student is in the office. If I seem to be busy with something else when you come by during office hours that’s just because no one is there yet and I’m looking for something else to do. Interrupt me!
- **Tutoring**. The Science Learning Resource Center (Sci 245, [http://www.smc.edu/AcademicPrograms/Tutoring/Pages/Science-LRC.aspx](http://www.smc.edu/AcademicPrograms/Tutoring/Pages/Science-LRC.aspx)), provides free tutoring for SMC students and other resources.
- **Suggested problems** for students to work from the textbook are listed online: [http://homepage.smc.edu/kline_peggy/chem-21/suggested-problems.html](http://homepage.smc.edu/kline_peggy/chem-21/suggested-problems.html). There is a strong correlation between working the problems in a serious and dedicated manner (pencil, paper, and lots of time) and course grade.
- **College Dates and Deadlines**. See the SMC Dates and Deadlines web page for enrollment and payment deadlines and Corsair Connect for individual course withdrawal deadlines The instructor reserves the right to drop any student who misses any class meetings during the first week. The instructor will probably drop students who miss a test without notification or appear to vanish; however, clerical errors do occur. If you want to be sure you are dropped, do it yourself. Aside from the circumstances under which you may be dropped by the instructor, it is nevertheless your responsibility as a student to withdraw from class if you do not intend to complete it. Students must not expect faculty to initiate withdrawal procedures for them. Students process a drop for themselves via Corsair Connect through 75% of the class, which is through the 12th week in a regular semester. Data regarding the withdrawal parameters for each class are provided within each student’s individual Corsair Connect account.
### Class Activities and Behavior

- **Successful completion** of this course will require full participation in all class activities. Punctuality is critical as well—plan to arrive on time each and every class period. You will miss important material, annoy your fellow students and anger the instructor when you disrupt the class by entering late. Students are responsible for knowing what happens in class, including schedule changes, material not in the book, information about what's going to be on the next test and so on. It is a good idea to have the names and contact information for a few students whom you can contact if you miss class.

- **Electronic devices.** Please adjust cell phones, laptops, tablets, etc. so they do not make noise and/or disrupt class members; the instructor reserves the right to confiscate such devices that do make noise and/or to evict students who are not using them appropriately during class time.

- **No eating, gum chewing, or drinking** is permitted in classrooms or labs; no food or drink is permitted unless it’s sealed so that it absolutely cannot spill. Water is allowed.

- **In-class points** will be for problems worked in groups during class time and/or for working problems at the board.

### Lab Information

- **See the Lab Notebooks and Reports handout** and the Chem 21 Lab webpage (http://homepage.smc.edu/kline_p Peggy/ chem-21-lab.html) for additional information about the lab and lab reports.

- **Turnitin.** This course will require electronic submission of written assignments to Turnitin via Canvas. Turnitin’s Originality Check conducts textual similarity reviews of submitted papers. When papers are submitted to Turnitin, the service may retain a copy of the submitted work in the Turnitin database for the sole purpose of detecting plagiarism in future submitted works. Students retain copyright on their original course work. Please note that Turnitin does not accuse you of plagiarism; it is only identifying similarity to content in its database, and you or I will determine whether you used source material accurately and ethically. Additionally I will use the GradeMark feature to grade your labs online. The use of Turnitin is subject to the Terms of Use agreement posted on the Turnitin.com website.

- **Personal Protective Equipment (PPE).** Chemical splash goggles must be worn by all students whenever they, or anyone else, are working with reagents in the laboratory. An extremely limited supply of goggles may be available in the lab for students to borrow. If you do not have a pair of appropriate goggles to wear for lab and none are available in the lab, you will have two options—go buy some or don’t work in lab that day. Students must supply nitrile or neoprene gloves for lab. Gloves must be removed and placed in the glove waste container when contaminated or any time you leave the lab. Students in organic chemistry classes must wear a blue flame-resistant lab coat whenever anyone in the room is conducting an experiment. If a student forgets his/her lab coat, he or she may not remain in the lab without it. A limited supply of lab coats is available to borrow.

- **A link to the SMC Safety Rules, PPE Information, and Emergency Information** is on the course web site. All students must sign a statement indicating that they are familiar with the this information before being allowed to work in the lab.

- **Lab neatness.** Students are responsible for keeping the lab neat. Students will be assigned one clean-up day during the semester. Any chemical samples kept at the end of the lab period must be labeled with the student's and instructor’s name, date, and identity of substance(s) present. Dispose of unknowns in the appropriate waste container, clean the vial in which they came, and return the empty clean vial to the instructor. There are ten points available for lab citizenship: deductions from the ten points will occur for transgressions such as lids not replaced on bottles, messy balance areas, not doing your assigned cleanup, not wearing appropriate PPE, etc. The entire lab class may get points deducted for certain items.

- **Missed labs.** There will be no make-up labs, unless you arrange to do the lab during another lab time when that section is doing the experiment. If you miss one lab, it will be the one dropped. If you miss more than one lab, you will forfeit the points. Students who must miss their regularly-scheduled lab time due to a day of religious observance may not be penalized; however, they must make a reasonable attempt to attend another session and/or complete the lab report. Students will not get credit for writing a lab report for a lab they did not attend.

- **Students who miss more than two labs** will not receive a passing grade in the course except in the case of exceptional circumstances.

- **Lab Reports** have two components: NCR copies of notebook pages that are turned in at various times and a typed portion that is submitted via Canvas/Turnitin. The latter is due at 11:59 pm, usually one or two weeks after the lab is finished. See Canvas for due dates. Each student may submit one lab lab late without penalty, if it is submitted within 24 hours of the original due date and time. See Lab Notebooks and Reports Handout for more information.

- **Lab Quizzes** are not dropped.
Accessibility and Accommodations

- This course is designed to be welcoming to, accessible by everyone, including students who are English-language learners, have a variety of learning styles, have disabilities, or are new to online learning. Be sure to let me know immediately if you encounter a required element or resource in the course that is not accessible to you. Also, let me know of changes I can make to the course so that it is more welcoming, accessible, or usable by students who take this course in the future.

- SMC accommodates students with disabilities. If you qualify for any special accommodations due to a disability, you must officially process your request through the Disabled Students Programs and Services (DSPS) office. Instructor must receive Testing Accommodation Authorization Form before the first anticipated accommodation and with sufficient notice to provide materials to DSPS. No retroactive accommodations will be provided. The student is solely responsible for securing any provisions to which they may be entitled. Scheduling of accommodated exams must be made through DSPS. Students must provide the instructor with the Testing Accommodation Appointment form before each test or quiz for which they plan to use DSPS and not change their mind. Be cognizant of DSPS hours. The DSPS office is located in the Admissions/Student Services Complex, Room 101, and the phone numbers are (310) 434-4265 and (310) 434-4273 (TDD). If you believe you have a learning disability that has not yet been documented, please make an appointment at the DSPS office.

- Religious Observance Absences. SMC AR 5530 states: “It is the college practice that students may be required to make-up missed work from absences due to the observance of a religious holiday, but they cannot be penalized for such absences. This practice applies to any work affecting a student’s grade.” Students must let the instructor know via email no later than the date given on the course Schedule of any planned absence due to a religious holiday. There is no guaranteed accommodation for notifications provided after the deadline. This course will avoid tests and quizzes on religious holidays that commonly affect students and, whenever possible, will schedule labs that can easily be rescheduled or done on a student’s own time on religious holidays that affect large numbers of students. Lectures will be held on religious holidays and students who miss class must get notes and information from other students.

Study Advice
Chemistry is not a spectator sport—reading the book and watching the instructor work problems will not be sufficient. The only way to learn to work problems is by working them yourself (without looking at the solutions). Read the appropriate chapter sections and work the problems as we cover them in lecture. Do not wait until the last minute to work the problems. It is often helpful at least to skim the chapter (read the headlines) before we cover it in lecture. If you have difficulty with a problem, try rereading the appropriate section(s) or your work the problems as we cover them in lecture. Do not wait until the last minute to work the problems. It is often helpful at least to skim the

Official Course Information

Link to official course outline - http://www.curricunet.com/SantaMonica/reports/course_outline_html.cfm?courses_id=2229

Catalogue Description: This course is a systematic introduction to the chemistry of carbon compounds. It encompasses theory and reactions of hydrocarbons and functional group derivatives. Included are bonding and structure, nomenclature, stereochemistry, synthesis, mechanism, and spectroscopic analysis. The laboratory work focuses on techniques of synthesis, isolation, purification, and analysis. Chem 21 and 22 together constitute two semesters of organic chemistry with one semester of organic chemistry laboratory. Chem 21, 22, and 24 constitute two semesters of organic chemistry with two semesters of laboratory. Prerequisite: Chemistry 12 or equivalent with a grade of C or better. Successful completion of this course requires access to a computer with internet access to complete homework assignments, download handouts, obtain lab information, and turn in lab reports.

Content: Review of Pertinent Chemistry 11 and 12 Topics—Bonding, Molecular Structure, Kinetics, Equilibrium, and Acid-Base Chemistry; Introduction to Organic Compounds—Functional Groups, Nomenclature, Reaction Concepts; Alkenes and their Reactions; Alkynes and their Reactions; Stereochemistry; Spectroscopy (MS, IR, NMR, uV-vis); Electron Delocalization and Resonance; Reactions of Dienes; Reactions of Alkanes; Nucleophilic Substitution and Elimination Reactions of Alkyl Halides; Alkane Reactions; Laboratory Experiments—Experiments utilizing common organic lab techniques such as reflux, distillation, extraction, recrystallization, melting point determination, chromatography, and spectroscopy in the preparation, isolation, purification, and identification of organic compounds.

Student Learning Outcomes As assessed by: questions on exams and/or observation of laboratory performance and/or evaluation of notebook data and lab reports and/or lab tests.

1. The student will follow a logical process based on well-established scientific principles and demonstrate the ability to use the appropriate problem-solving techniques to solve a scientific problem such as an organic synthesis comprised of two or more steps, or a determination of the structure of a compound based on spectroscopy (IR, NMR, MS) and/or chemical evidence.

2. When conducting a laboratory experiment, the student will follow written procedures commonly used in the organic lab (such as reflux, distillation, extraction, recrystallization, and melting-point determination) accurately and safely. When completing a lab report, the student will apply the scientific method correctly by being able to state a hypothesis, take careful measurements, estimate uncertainties and draw appropriate conclusions based on gathered data and scientific principles.

3. The student will explain observable phenomena using appropriate scientific theories, such as writing a reaction mechanism consistent with observed facts or utilizing appropriate structures to represent different types of organic molecules and species.